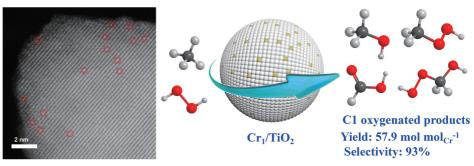
## Single Chromium Atoms for Direct Methane Conversion

ethane is one of the most promising building blocks for producing basic chemicals because of it is abundant and inexpensive. However, methane is the most inert hydrocarbon owing to the strong C-H bond strength (434 kJ mol<sup>-1</sup>) and its tetrahedral structure. Activation of methane is a big challenge

and has been regarded as the holy grail in catalysis.

Recently, a joint research group from the Institute of Chemistry of Chinese Academy of Sciences (ICCAS), led by Prof. SONG Weiguo, reported single chromium atoms supported on titanium dioxide nanoparticles are an efficient heterogeneous catalyst for direct methane oxidation to C1 oxygenated products with H<sub>2</sub>O<sub>2</sub> as oxidant under mild conditions. The work entitled "Single Chromium Atoms Supported on Titanium Dioxide Nanoparticles for Synergic Catalytic Methane Conversion under Mild Conditions" was published in Angew. Chem. Int. Ed.

With optimized catalyst, the highest yield for C1 oxygenated products of 57.9 mol mol<sub>C</sub><sup>-1</sup> can be reached with a high selectivity of around 93% at 50°C for 20 h, which is significantly higher than those of most reported catalysts. The superior catalytic performance can be attributed to the synergistic effect between Cr single



HAADF-STEM image of Cr<sub>1</sub>/TiO<sub>2</sub> and its catalytic property for CH<sub>4</sub> oxidation. (Image by Prof. SONG Weiguo)

atoms and TiO<sub>2</sub> support. Combining catalytic kinetics, electron paramagnetic resonance and control experiment results, a possible reaction pathway for methane conversion with Cr<sub>1</sub>/TiO<sub>2</sub> catalyst is proposed as: CH4 is first oxidized through a radical pathway to form CH<sub>3</sub>OH and CH<sub>3</sub>OOH, and then the generated CH<sub>3</sub>OH is further oxidized to HOCH<sub>2</sub>OOH and HCOOH. This study provides a new non-noble metal single-site catalyst with superior catalytic performance for direct conversion of methane under mild conditions.

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(ICCAS)

## Contact:

Prof. SONG Weiguo Institute of Chemistry, CAS Email: wsong@iccas.ac.cn

## Reference

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